# nannvi

# Oxalyl Amide Assisted Palladium-Catalyzed Arylation of C(sp<sup>2</sup>)–H Bond at the  $\delta$  Position

Jian Han, Pei Liu, Chao Wang, Qian Wang, Jingyu Zhang, Yanwei Zhao, Daqing Shi, Zhibin Huang,\* and Yingsheng Zhao\*

Key Laboratory of Organ[ic](#page-3-0) Synthesis of Jiangsu Province, College of Chemistry, Chemical Engineering and Materials Science, Soochow University, Suzhou 215123, China

**S** Supporting Information



ABSTRACT: A successful protocol has been developed for  $\delta$ -arylation of  $\beta$ -arylethamines at the ortho position under mild conditions. The newly developed methodology first presents broad substrate scope, great functional group tolerance, and good to excellent yield in the synthesis of substituted β-arylethylamines. The transformation represents a practical advantage of oxalyl amide in assistance with C−H functionalization at a remote position.

Transition-metal-catalyzed C<sup>−</sup>H bond activation for C−<sup>C</sup> bond formation reactions has regained considerable attention in the last decades and has become an efficient method for synthesis of natural products and pharmaceutical agents.<sup>1</sup> Among these reports, functionalization of directing-groupco[n](#page-3-0)taining arenes or unactivated  $C(sp^3)$ -H bonds has been extensively investigated.<sup>2</sup> Especially in recent years, several groups have demonstrated that amide groups possess a superior ability to assist palladiu[m-](#page-3-0)catalyzed regioselective functionalization of  $C(sp^2)$ –H or  $C(sp^3)$ –H bonds.<sup>3</sup> However, only a few directing groups derived from amine derivatives were reported (Scheme 1A). For example, Daugulis and co-workers first discovered palladium-catalyzed γ-arylation of amine derivatives by applying picolinamide as directing group. Later, Chen and co-

# Scheme 1. Protecting Group Directed C−H Activation

A) Protecting group directed C-H activation for amines



workers reported the picolinamide-assisted γ-functionalization of natural amino acids in moderate yields.<sup>4</sup> In 2013,  $N-(2$ pyridyl)sulfonyl-promoted γ-arylation of amino acid derivatives at a high reaction temperature was reported [by](#page-3-0) Carretero and coworkers.<sup>5</sup> Ma and Fan also developed a protocol for *γ*-arylation of substituted 2-aminobutanates by employing 2-methoxyiminoacetyl as the directing group with limited substrate scope.<sup>6</sup> Interestingly, the palladium-catalyzed ortho trifluoromethylation of N-unsubstituted benzyl amines could be directly achieve[d,](#page-3-0) which was discoverd by the Yu group.<sup>7</sup> Very recently, Yu and coworkers also developed γ-arylation of C(sp<sup>3</sup>)−H bonds of triflylprotected amines with arylboron reag[e](#page-3-0)nts through application of mono-N-protected amino acid as a ligand.<sup>8</sup>

Although these directing groups had made a wide variety of transformations and greatly enriched the [r](#page-3-0)eaction scope, these auxilixary-promoted palladium-catalyzed C−C bond formation reactions mostly go through a kinetically favored five-membered palladcycle intermediate to achieve  $\gamma$ -arylation.<sup>9</sup> To date, there are few reports of directing group assisted selective  $\delta$ - and  $\varepsilon$ arylation of amine derivatives which undergo [a](#page-3-0) six- or sevenmembered palladacycle pathway.<sup>10</sup> Herein, we reported an efficient method for the arylation of  $\delta$ -C(sp<sup>2</sup>)–H bonds of oxalyl amide protected  $\beta$ -arylethamine[s w](#page-3-0)ith aryl iodides via sixmembered palladcycles. Our new methodology showed a wide substrate scope and great functional group tolerance in the synthesis of substituted  $\beta$ -arylethylamines under mild conditions, which disclosed the super assistance ability of oxalyl amide at remote position in C−H functionalization (Scheme 1B).

The weak coordinating directing groups, such as carboxylic acid,<sup>11</sup> hydroxyl,<sup>12</sup> methoxyl,<sup>13</sup> and ketone<sup>14</sup> could selectively achieved different types of C−H functionalization. Generally,

Received: September 17, 2014 Published: October 16, 2014





 $a^a$ Reaction conditions: 1a–1 (0.2 mmol), p-Tol-I (0.3 mmol),  $Pd(OAc)_{2}$  (2.5 mol %),  $K_{2}CO_{3}$  (0.4 mmol), PivOH (0.06 mmol), DCE (0.3 mL), 80 °C, 24 h. Isolated yields.

these weak coordinating auxiliaries promoted  $\rm C(sp^2)-H$ activation could selectively happened at  $\delta$  positions. The  $\delta$ alkenylation of the triflyl-protected  $\beta$ -arylethamine was also developed by the Yu group in 2008.<sup>15</sup> Very recently, oxalyl amide was discovered as an easily accessible and efficient directing group<sup>16</sup> for selective C−H activati[on](#page-3-0) at  $\delta$  or  $\varepsilon$  position by our group. Inspired by the assistance ability of oxalyl amide, we hypot[he](#page-3-0)sized that the new developed N,O-bidentated directing group for amine derivatives might promote the selective arylation of C(sp<sup>2</sup>)–H bonds at  $\delta$  or  $\varepsilon$  position.

The phenylethylamine derivatives are important versatile sythetic precursors.<sup>17</sup> We first tested acetylated, triflated, and picolinamide-protected phenylethylamine to realize arylation at ortho position und[er](#page-3-0) various conditions. However, all failed to give the arylated product in acceptable yield under mild conditions (2a−c). We were pleased to find that the desired arylated product was obtained in 15% yield by employing methyl oxalyl ester (2d) as directing group, following Fagnou's base system,<sup>18</sup> using  $Pd(OAc)_2$  (0.025 equiv), 4-iodotoluene (1.5 equiv),  $K_2CO_3$  (2 equiv), PivOH (0.3 equiv), and DCE as solvent [a](#page-3-0)t 80 °C. The catalytic amount of pivalic acid was necessary for the transformation (for detailed optimal reaction conditions, see the Supporting Information). It is suggested that the pivalate anion acts as a proton shuttle between substrate and inorganic base as [disclosed by Larock and](#page-2-0) Fagnou.<sup>18</sup> It is not surprising that the ortho-arylated product 2e could be isolated in 72% yield when we applied oxalyl amide instead [of](#page-3-0) ester. To

Table 2. Scope of Aryl Iodides Reagents for  $\delta\text{-C(sp}^2\text{)-H}$ Arylation Reaction<sup>a</sup>



<sup>a</sup>Reaction conditions: 1g (0.2 mmol), ArI (0.3 mmol), Pd(OAc)<sub>2</sub> (2.5 mol %), K<sub>2</sub>CO<sub>3</sub> (0.4 mmol), PivOH (0.06 mmol), DCE (0.3 mL), 80<br>°C, 24 h. Isolated yields. <sup>b</sup>100 °C. <sup>c</sup>Ag<sub>2</sub>CO<sub>3</sub> (0.2 mmol) was used instead of  $K_2CO_3$ , 100 °C.  ${}^{d}Pd(OAc)_2$  (5 mol %), 100 °C.

further understand the assistance ability of different amides, a series of oxalyl amide protected phenylethylamines were prepared and screened (Table 1, 2e−l). Generally, much better yields were achieved when we applied oxalyl amide instead of ester (2e−g). Diethylamine could increase the yield up to 81% (2f). To our delight, compounds 1g give the corresponding arylated product in best yield  $(2g)$ . Further, when more bulky secondary amine was applied, the yield dramatically decreased (2h,i). Piperidine and morpholine were less active, and more than 40% of starting material was recovered, respectively  $(2j,k)$ . tert-Butyl amine directly shut down the reaction as the strong coordination to the palladium center (2l). Thus, we chose the oxalyl amide, which was made from general reagents of diisopropylamine and oxalyl chloride as the directing group to test the scope of primary amines as arylation substrates.

Using the newly developed protocol, a broad range of aryl iodides were successfully applied with substrate 1g as shown in Table 2. Gratifyingly, the electron-withdrawing groups such as Br-, I-, Ac-, CN-, NHAc-, CHO-,  $NO<sub>2</sub>$ -, and CF<sub>3</sub>-substituted benzene iodides were tolerated in this transformation and gave good to excellent yields. It is worth mentioning that even the iodine substitution on the aromatic ring was achieved in 71% yield by increasing the reaction temperature to 100  $^{\circ}$ C (3f). The electron-rich aryl iodide also showed nice reactivity and gave excellent yield (3k). To our interest, the pyridine substrate also gave the arylated product 3l in 67% yield employing 5 mol % of

<span id="page-2-0"></span>

 $a^a$ Reaction conditions: 4a−j (0.2 mmol), p-Tol-I (0.3 mmol),  $Pd(OAc)_2$  (2.5 mol %),  $K_2CO_3$  (0.4 mmol), PivOH (0.06 mmol), DCE (0.3 mL), 80 °C, 24 h. Isolated yields.  $b^{b}100$  °C. <sup>c</sup>120 °C. <sup>d</sup>4k–m (0.2 mmol), p-Tol-I (0.6 mmol), Pd(OAc)<sub>2</sub> (5 mol %), Ag<sub>2</sub>CO<sub>3</sub> (0.3) mmol),  $120^{\circ}$ C.  $^{e}$ **4n** (0.2 mmol),  $p$ -Tol-I (0.3 mmol),  $Pd(OAc)_{2}$  (5 mol %), Ag<sub>2</sub>CO<sub>3</sub> (0.3 mmol), 100 °C.





#### Scheme 3. Preliminary Mechanistic Study

A) Deuterium experiments



palladium acetate at 100 °C. However, when we tried to use 4 bromotoluene, just 15% yield of desired product was isolated (Supporting Information).

We next moved our attention to other  $\beta$ -arylethyamines. Good yields and high selectivity were observed for all of the substrates under mild conditions (Table 3). The electronic factor had nearly no effect on these arylation reactions at the  $\delta$  position (5a−i) except for the substrate (4f). However, substrates 4k−m gave mono- and diarylated products without any selectivity when  $K_2CO_3$  (2 equiv), PivOH (0.3 equiv), and 4-iodotoluene (3 equiv) were used under palladium catalyst. When we applied silver carbonate instead of potassium carbonate and pivalic acid, satisfactory yields were obtained (5k−m), indicating the better reactivity of silver carbonate than that of potassium carbonate and pivalic acid catalyst system. High monoarylation took place regioselectively at the less hindered site  $(5b,c,f,i,j)$ . To the best of our knowledge, selective arylation of amine derivatives at  $\varepsilon$ position still has no reports. Delightedly, the substrate of 4n could be arylated at the  $\varepsilon$  position in moderate yield via palladium catalyst (5n).

The gram-scale reaction was successfully achieved with 0.5 mol % of  $Pd(OAc)_2$  in 80% yield, which highlighted the power of this approach (Scheme 2). The easily accessible and removable ability was crucial for auxiliary in C−H activation.<sup>19</sup> Notably, the new developed directing group fulfilled these requirements. The oxalyl amide could be easily prepared from diisop[rop](#page-3-0)ylamine and oxalyl chloride which are cheap and easily accessible reagents. Besides, the oxalyl amide could be removed readily under mild conditions in excellent yield (Scheme 2).

The ortho C−H bond of substrate 4h was deuterated under the catalysis of  $Pd(OAc)$ , in AcOD, giving compound  $[D]$ -4h in 92% yield.<sup>20</sup> A primary kinetic isotope effect ( $\sim$ 2.9) was observed under the general arylation conditions, implicating C−H activation [a](#page-3-0)s the rate-determing step in the catalytic cycle (Supporting Information). $21$ 

To further investigate the mechanism of arylation promoted by oxalyl amide, substrate [4h](#page-3-0) with 2 equiv of acetonitrile and 1 equiv of Pd(OAc)<sub>2</sub> in benzene- $d_6$  at 80 °C was monitored at different times. The results indicated that C−H activation can take place without involvement of the ArI substrates (see the Supporting Information). We then tried to get the six-membered palladacycle intermediate in toluene and acetonitrile. Unfortunately, we failed to obtain the crystal of palladacycle intermediate. Based on our primary study and recent reports,  $\rm ^{4b,22}$  we proposed the mechanism might go through a concerted metalation− deprotonation (CMD) pathway to give the co[mple](#page-3-0)x 8 and then an ArI-coupling pathway for the C−H arylation (Scheme 3).

In conclusion, we have developed an easily accessible and highly effective directing group for functionalization of C−H bonds at the  $\delta$  and  $\varepsilon$  positions. Arylation of the  $\beta$ -arylethamines at the ortho position was initially studied and gave good to excellent yields by using palladium as catalyst under mild conditions. Detailed mechanistic studies are in progress, and study of a new application of oxalyl amide as a directing group in the construction of C−C, C−F, or C−heteroatom bonds is underway in our laboratory.

# **ASSOCIATED CONTENT**

#### S Supporting Information

Experimental procedures and spectroscopic data for all new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

# <span id="page-3-0"></span>■ AUTHOR INFORMATION

## Corresponding Authors

\*E-mail: zbhuang@suda.edu.cn. \*E-mail: yszhao@suda.edu.cn.

### Notes

The authors declare no competing financial interest.

# ■ ACKNOWLEDGMENTS

This research was financially supported by Natural Science Foundation of Jiangsu Province of China (L210903913), a startup fund (Q410901212) from Soochow University, and the Young National Natural Science Foundation of China (NO.21402133) for support of this work. The PAPD is also gratefully acknowledged.

# ■ REFERENCES

(1) For recent reviews: (a) Giri, R.; Shi, B.-F.; Engle, K. M.; Maugel, N.; Yu, J.-Q. Chem. Soc. Rev. 2009, 38, 3242. (b) Lyons, T. W.; Sanford, M. S. Chem. Rev. 2010, 110, 1147. (c) Colby, D. A.; Bergman, R. G.; Ellman, J. A. Chem. Rev. 2010, 110, 624. (d) McMurray, L.; O'Hara, F.; Gaunt, M. J. Chem. Soc. Rev. 2011, 40, 1885. (e) Li, B.-J.; Shi, Z.-J. Chem. Soc. Rev. 2012, 41, 5588. (f) Mousseau, J. J.; Charette, A. B. Acc. Chem. Res. 2012, 46, 412. (g) Louillat, M.-L.; Patureau, F. W. Chem. Soc. Rev. 2014, 43, 901.

(2) For examples, see: (a) Colby, D. A.; Bergman, R. G.; Ellman, J. A. Chem. Rev. 2010, 110, 624. (b) Engle, K. M.; Mei, T.-S.; Wasa, M.; Yu, J.- Q. Acc. Chem. Res. 2011, 45, 788. (c) Colby, D. A.; Tsai, A. S.; Bergman, R. G.; Ellman, J. A. Acc. Chem. Res. 2011, 45, 814. (d) Yeung, C. S.; Dong, V. M. Chem. Rev. 2011, 111, 1215. (e) Roizen, J. L.; Harvey, M. E.; Bois, J. Du. Acc. Chem. Res. 2012, 45, 911. (f) Ackermann, L. Acc. Chem. Res. 2014, 47, 281.

(3) Examples for directing group assisted C−H activation: (a) Wasa, M.; Engle, K. M.; Yu, J.-Q. J. Am. Chem. Soc. 2009, 131, 9886. (b) Daugulis, O.; Do, H.-Q.; Shabashov, D. Acc. Chem. Res. 2009, 42, 1074. (c) Romero-Revilla, J. A.; García-Rubia; Goméz Arrayás, A. R.; Fernández-Ibáñez, M. Á.; Carretero, J. C. *J. Org. Chem.* **2011**, 76, 9525. (d) Cho, S. H.; Kim, J. Y.; Kwak, J.; Chang, S. Chem. Soc. Rev. 2011, 40, 5068. (e) Park, S. H.; Kim, J. Y.; Chang, S. Org. Lett. 2011, 13, 2372. (f) Schröder, N.; Wencel-Delord, J.; Glorius, F. J. Am. Chem. Soc. 2012, 134, 8298. (g) Aihara, Y.; Chatani, N. J. Am. Chem. Soc. 2013, 136, 898. (h) Zhang, Q.; Chen, K.; Rao, W.; Zhang, Y.; Chen, F.-J.; Shi, B.-F. Angew. Chem., Int. Ed. 2013, 52, 13588.

(4) The examples for picolinic amide as directing group: (a) Zaitsev, V. G.; Shabashov, D.; Daugulis, O. J. Am. Chem. Soc. 2005, 127, 13154. (b) Shabashov, D.; Daugulis, O. J. Am. Chem. Soc. 2010, 132, 3965. (c) Feng, Y.; Wang, Y.; Landgraf, B.; Liu, S.; Chen, G.Org. Lett. 2010, 12, 3414. (d) Zhao, Y.; Chen, G. Org. Lett. 2011, 13, 4850. (e) He, G.; Lu, C.; Zhao, Y.; Nack, W. A.; Chen, G. Org. Lett. 2012, 14, 2944. (f) Zhang, S.-Y.; Li, Q.; He, G.; Nack, W. A.; Chen, G. J. Am. Chem. Soc. 2013, 135, 2124.

(5) Examples for the N-(2-pyridyl)sulfonyl as directing group: (a) García-Rubia, A.; Arrayás, R. G.; Carretero, J. C. Angew. Chem., Int. Ed. 2009, 48, 6511. (b) Rodriguez, N.; Romero-Revilla, J. A.; Fernandez-Ibanez, M. A.; Carretero, J. C. Chem. Sci. 2013, 4, 175. (6) Fan, M.; Ma, D. Angew. Chem., Int. Ed. 2013, 52, 12152.

(7) Miura, M.; Feng, C.-G.; Ma, S.; Yu, J.-Q. Org. Lett. 2013, 15, 5258. (8) The examples for triflamid as directing group: (a) Li, J.-J.; Mei, T.- S.; Yu, J.-Q. Angew. Chem., Int. Ed. 2008, 47, 6452. (b) Mei, T.-S.; Wang, X.; Yu, J.-Q. J. Am. Chem. Soc. 2009, 131, 10806. (c) Vickers, C. J.; Mei, T.-S.; Yu, J.-Q. Org. Lett. 2010, 12, 2511.

(9) (a) Alsters, P. L.; Engel, P. F.; Hogerheide, M. P.; Copijn, M.; Spek, A. L.; van Koten, G. Organometallics 1993, 12, 1831. (b) Dupont, J.; Consorti, C. S.; Spencer, J. Chem. Rev. 2005, 105, 2527. (c) Deprez, N. R.; Sanford, M. S. J. Am. Chem. Soc. 2009, 131, 11234.

(10) (a) Li, J.-J.; Mei, T.-S.; Yu, J.-Q. Angew. Chem., Int. Ed. 2008, 47, 6452. (b) Yang, Y.-F.; Cheng, G.-J.; Liu, P.; Leow, D.; Sun, T.-Y.; Chen, P.; Zhang, X.; Yu, J.-Q.; Wu, Y.-D.; Houk, K. N. J. Am. Chem. Soc. 2013, 136, 344. (c) Li, J.-J.; Giri, R.; Yu, J.-Q. Tetrahedron 2008, 64, 6979.

(11) Examples for carboxylic acid as directing group: (a) Giri, R.; Yu, J.- Q. J. Am. Chem. Soc. 2008, 130, 14082. (b) Wang, D.-H.; Engle, K. M.; Shi, B.-F.; Yu, J.-Q. Science 2010, 327, 315.

(12) Examples for hydroxyl as directing group: (a) Lu, Y.; Wang, D.-H.; Engle, K. M.; Yu, J.-Q. J. Am. Chem. Soc. 2010, 132, 5916. (b) Wang, X.; Lu, Y.; Dai, H.-X.; Yu, J.-Q. J. Am. Chem. Soc. 2010, 132, 12203. (c) Xiao, B.; Gong, T.-J.; Liu, Z.-J.; Liu, J.-H.; Luo, D.-F.; Xu, J.; Liu, L. J. Am. Chem. Soc. 2011, 133, 9250.

(13) The examples for methoxyl as directing group: (a) Li, G.; Leow, D.; Wan, L.; Yu, J. Q. Angew. Chem., Int. Ed. 2012, 52, 1245. (b) Iglesias, Á .; Á lvarez, R.; de Lera, Á . R.; Muñiz, K. Angew. Chem., Int. Ed. 2012, 51, 2225.

(14) (a) Murai, S.; Kakiuchi, F.; Sekine, S.; Tanaka, Y.; Kamatani, A.; Sonoda, M.; Chatani, N. Nature 1993, 366, 529. (b) Kakiuchi, F.; Murai, S. Acc. Chem. Res. 2002, 35, 826. (c) Xiao, B.; Fu, Y.; Xu, J.; Gong, T.-J.; Dai, J.-J.; Yi, J.; Liu, L. J. Am. Chem. Soc. 2009, 132, 468.

(15) Li, J. J.; Mei, T. S.; Yu, J. Q. Angew. Chem., Int. Ed. 2008, 47, 6452. (16) (a) Wang, C.; Chen, C.; Zhang, J.; Han, J.; Wang, Q.; Guo, K.; Liu, P.; Guan, M.; Yao, Y.; Zhao, Y. Angew. Chem., Int. Ed. 2014, 53, 9884. (b) Wang, Q.; Han, J.; Wang, C.; Zhang, J.; Huang, Z.; Shi, D.; Zhao, Y. Chem. Sci. 2014, DOI: 10.1039/c4sc02172j.

(17) Sahakitpichan, P.; Ruchirawat, S. Tetrahedron Lett. 2003, 44, 5239. (18) (a) Lafrance, M.; Fagnou, K. J. Am. Chem. Soc. 2006, 128, 16496. (b) Huang, Q.; Fazio, A.; Dai, G.; Campo, M. A.; Larock, R. C. J. Am. Chem. Soc. 2004, 126, 7460.

(19) Ye, X.; He, Z.; Ahmed, T.; Weise, K.; Akhmedov, N. G.; Petersen, J. L.; Shi, X. Chem. Sci. 2013, 4, 3712.

(20) Ma, S.; Villa, G.; Thuy-Boun, P. S.; Homs, A.; Yu, J.-Q. Angew. Chem., Int. Ed. 2014, 53, 734.

(21) (a) Desai, L. V.; Stowers, K. J.; Sanford, M. S. J. Am. Chem. Soc. 2008, 130, 13285. (b) Deprez, N. R.; Sanford, M. S. J. Am. Chem. Soc. 2009, 131, 11234.

(22) (a) Sun, C. L.; Li, B. J.; Shi, Z.-J. Chem. Commun. 2010, 46, 677. (b) Zhao, X.; Yeung, C. S.; Dong, V. M. J. Am. Chem. Soc. 2010, 132, 5837.